

CHAPTER 1 NATURE OF LIGHT

1-1. a) $\lambda = \frac{h}{p} = \frac{h}{m v} = \frac{6.63 \times 10^{-34} \text{ J} \cdot \text{s}}{(0.05 \text{ kg})(20 \text{ m/s})} = 6.63 \times 10^{-34} \text{ m}$

b) $\lambda = \frac{h}{p} = \frac{h}{\sqrt{2 m E}} = \frac{6.63 \times 10^{-34} \text{ J} \cdot \text{s}}{[(2 \cdot 9.11 \times 10^{-31} \text{ kg})(10 \cdot 1.602 \times 10^{-19} \text{ J})]^{1/2}} = 3.88 \times 10^{-10} \text{ m}$

1-2. $P = \frac{\text{Energy}}{\text{time}} = \frac{n h \nu}{t} = \frac{n h c}{t \lambda} = \frac{100 (6.63 \times 10^{-34} \text{ J} \cdot \text{s}) (3 \times 10^8 \text{ m/s})}{(1 \text{ s})(550 \times 10^{-9} \text{ m})} = 3.62 \times 10^{-17} \text{ W}$

1-3. The energy of a photon is given by $E = h \nu = h c / \lambda$

At $\lambda = 380 \text{ nm}$: $E = \frac{(6.63 \times 10^{-34} \text{ J} \cdot \text{s}) (3 \times 10^8 \text{ m/s})}{380 \times 10^{-9} \text{ m}} = (5.23 \times 10^{-19} \text{ J}) \frac{1 \text{ eV}}{1.60 \times 10^{-19} \text{ J}} = 3.27 \text{ eV}$

At $\lambda = 770 \text{ nm}$: $E = \frac{(6.63 \times 10^{-34} \text{ J} \cdot \text{s}) (3 \times 10^8 \text{ m/s})}{770 \times 10^{-9} \text{ m}} = (2.58 \times 10^{-19} \text{ J}) \frac{1 \text{ eV}}{1.60 \times 10^{-19} \text{ J}} = 1.61 \text{ eV}$

1-4. $p = E/c = m c^2/c = m c = 2.73 \times 10^{-22} \text{ kg} \cdot \text{m/s}$, $\lambda = \frac{h}{p} = \frac{h c}{E} = \frac{h c}{m c^2} = \frac{h}{m c} = 2.43 \times 10^{-12} \text{ m}$

1-5. $E_{v=0} = m c^2 = (9.109 \times 10^{-31} \text{ kg}) (2.998 \times 10^8 \text{ m/s})^2 = (8.187 \times 10^{-14} \text{ J}) \frac{1 \text{ MeV}}{1.602 \times 10^{-19} \text{ J}} = .511 \text{ MeV}$

1-6. $c p = \sqrt{E^2 - m^2 c^4}$, where $E = E_K + m c^2 = (1 + 0.511) \text{ MeV}$. So $c p = \sqrt{1.511^2 - 0.511^2} \text{ MeV}$

That is, $c p = 1.422 \text{ MeV}$ and $p = 1.422 \text{ MeV}/c$.

1-7. $\lambda = \frac{h c}{E} = \frac{(6.626 \times 10^{-34} \text{ J} \cdot \text{s}) (2.998 \times 10^8 \text{ m/s})}{E} \left(\frac{1 \text{ eV}}{1.602 \times 10^{-19} \text{ J}} \right) \left(\frac{1 \text{ \AA}}{10^{-10} \text{ m}} \right) = \frac{12,400}{E} (\text{\AA} \cdot \text{eV})$

1-8. $E_K = m c^2 \left(\frac{1}{\sqrt{1 - v^2/c^2}} - 1 \right) = m c^2 \left[(1 - v^2/c^2)^{-1/2} - 1 \right] \simeq m c^2 \left[(1 - (-1/2) v^2/c^2) - 1 \right] = \frac{1}{2} m v^2$

1-9. The total energy of the proton is,

$$E = E_K + m_p c^2 = 2 \times 10^9 \left(\frac{1.60 \times 10^{-19} \text{ J}}{1 \text{ eV}} \right) + (1.67 \times 10^{-27} \text{ kg}) (3.00 \times 10^8 \text{ m/s})^2 = 4.71 \times 10^{-10} \text{ J}$$

$$\text{a) } p = \frac{\sqrt{E^2 - m_p^2 c^4}}{c} = \frac{\left[(4.71 \times 10^{-10} \text{ J})^2 - (1.67 \times 10^{-27} \text{ kg})^2 (3.00 \times 10^8 \text{ m/s})^2 \right]^{-1/2}}{3.00 \times 10^8 \text{ m/s}}$$

$$p = 1.49 \times 10^{-18} \text{ kg} \cdot \text{m/s}$$

$$\text{b) } \lambda = h/p = (6.63 \times 10^{-34} \text{ J} \cdot \text{s}) / (1.49 \times 10^{-18} \text{ kg} \cdot \text{m/s}) = 4.45 \times 10^{-16} \text{ m}$$

$$\text{c) } \lambda_{\text{photon}} = h c/E = (6.63 \times 10^{-34} \text{ J} \cdot \text{s}) (3.00 \times 10^8 \text{ m/s}) / (4.71 \times 10^{-10}) = 4.22 \times 10^{-16} \text{ m}$$

$$\text{1-10. } n_{\text{photons}} = \frac{\text{Energy}}{h \nu} = \frac{\text{Energy}}{h c/\lambda} = \frac{(1000 \text{ W/m}^2) (10^{-4} \text{ m}^2)}{(6.63 \times 10^{-34} \text{ J}) (3.00 \times 10^8 \text{ m/s}) / (550 \times 10^{-9} \text{ m})} = 2.77 \times 10^{17}$$

$$\text{1-11. } \frac{n_1}{n_2} = \frac{E_e/h \nu_1}{E_e/h \nu_2} = \frac{E_e \lambda_1/h c}{E_e \lambda_2/h c} = \frac{\lambda_1}{\lambda_2}$$

1-12. The wavelength range is 380 nm to 770 nm. The corresponding frequencies are

$$\nu_{770} = \frac{c}{\lambda} = \frac{3.00 \times 10^8 \text{ m/s}}{770 \times 10^{-9} \text{ m}} = 3.89 \times 10^{14} \text{ Hz} \quad \nu_{380} = \frac{c}{\lambda} = \frac{3.00 \times 10^8 \text{ m/s}}{380 \times 10^{-9} \text{ m}} = 7.89 \times 10^{14} \text{ Hz}$$

1-13. The wavelength of the radio waves is $\lambda = c/\nu = (3.00 \times 10^8 \text{ m/s}) / (100 \times 10^6 \text{ Hz}) = 3 \text{ m}$. The length of the half-wave antenna is then $\lambda/2 = 1.5 \text{ m}$.

1-14. The wavelength is $\lambda = c/\nu = (3.0 \times 10^8 \text{ m/s}) / (90 \times 10^6 \text{ Hz}) = 3.33 \text{ m}$. The length of each of the rods is then $\lambda/4 = 0.83 \text{ m}$.

1-15. a) $t = D_l/c = (90 \times 10^3 / 3.0 \times 10^8) \text{ s} = 3.0 \times 10^{-4} \text{ s}$. b) $D_s = v_s t = (340) (3.0 \times 10^{-4}) \text{ m} = 0.10 \text{ m}$

1-16. a) $I_e = \frac{\Phi_e}{\Delta\omega} = \frac{500 \text{ W}}{4 \pi \text{ sr}} = 39.8 \text{ W/sr}$ b) $M_e = \frac{\Phi_e}{A} = \frac{500 \text{ W}}{5 \times 10^{-4} \text{ m}^2} = 10^6 \text{ W/m}^2$

$$\text{c) } E_e = \frac{\Phi_e}{A} = \frac{\Phi_e}{4 \pi r^2} = \frac{500 \text{ W}}{4 \pi (2 \text{ m})^2} = 9.95 \text{ W/m}^2 \quad \text{e) } \Phi_e = E_e A = (9.95 \text{ W/m}^2) \pi (0.025 \text{ m})^2 = .0195 \text{ W}$$

1-17. a) The half-angle divergence $\theta_{1/2}$ can be found from the relation

$$\tan(\theta_{1/2}) \approx \theta_{1/2} = \frac{r_{\text{spot}}}{L_{\text{room}}} = \frac{0.0025 \text{ m}}{15 \text{ m}} = 1.67 \times 10^{-4} \text{ rad} = .0096^\circ$$

$$\text{b) The solid angle is } \Delta\omega = \frac{A_{\text{spot}}}{L_{\text{room}}^2} = \frac{\pi r_{\text{spot}}^2}{L_{\text{room}}^2} = \frac{\pi (0.0025 \text{ m})^2}{(15 \text{ m})^2} = 8.73 \times 10^{-8} \text{ sr}.$$

$$\text{c) The irradiance on the wall is } E_e = \frac{\Phi_e}{A_{\text{spot}}} = \frac{\Phi_e}{\pi r_{\text{spot}}^2} = \frac{0.0015 \text{ W}}{\pi (0.0025 \text{ m})^2} = 76.4 \text{ W/m}^2.$$

d) The radiance is (approximating differentials as increments)

$$L_e \approx \frac{\Phi_e}{\Delta\omega \Delta A_{\text{laser}} \cos\theta} = \frac{0.0015 \text{ W}}{(8.73 \times 10^{-8} \text{ sr}) (\pi (0.00025 \text{ m})^2) \cos(0)} = 8.75 \times 10^{10} \frac{\text{W}}{\text{m}^2 \cdot \text{sr}}$$